

## ABIETANE DITERPENOIDS OF *SALVIA ANASTOMOSANS*\*

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**Abstract**—The aerial parts of *Salvia anastomosans* afforded, in addition to oleanolic and ursolic acids, three abietane diterpenoids: conacytone and icetexone, of known structures, and anastomosine, a new 9(10→20)-abeo-abietane diterpenoid whose structure was elucidated by spectroscopic and chemical means.

### INTRODUCTION

The genus *Salvia* L., is represented in Mexico by over 300 species classified in the subgenus *Calosphace* [1, 2]. Our systematic study of the Mexican *Salvia* species has shown that the diterpenoid content of plants studied depends on the Section to which it belongs [3]. Most of the diterpenoids isolated from *Salvia* species, subgenus *Calosphace*, are neo-clerodane diterpenoids or can be derived biogenetically from a clerodane precursor [3–5]. The presence of abietane diterpenoids in the species of the Section *Erythrostachys* [6] and 9(10→20)-abeo-abietane diterpenoids in the species of the Section *Tomentellae* studied up to now [7, 8], could have a phytogeographical significance.

In continuation of our systematic studies of Mexican *Salvia* spp., we have analysed the aerial parts of *S. anastomosans* Ramamoorthy, a perennial shrub recently classified in the Section *Tomentellae* (*Salvia*, subgenus *Calosphace*)[9].

### RESULTS AND DISCUSSION

Extraction of the aerial parts of a population of *S. anastomosans* collected near Tamazulapán (Oaxaca) afforded, after extensive chromatography, oleanolic and ursolic acids and three abietane diterpenes, conacytone (1) and icetexone (2) previously isolated from *S. ballotaeflora*, and anastomosine to which we assigned structure 3.

Anastomosine (3) has the molecular formula  $C_{20}H_{20}O_5$ . Its IR spectrum exhibited the characteristic absorptions due to a chelated hydroxyl group ( $3361\text{ cm}^{-1}$ ), *p*-quinoid carbonyls ( $1656\text{ cm}^{-1}$ ), double bonds ( $1602\text{ cm}^{-1}$ ) and a saturated  $\gamma$ -lactone function ( $1778\text{ cm}^{-1}$ ).

The  $^1\text{H}$  NMR spectrum (Table 1) showed the signals due to an isopropyl group attached to a *p*-quinone ring ( $\delta 3.38$ , 1H, septet,  $J = 7\text{ Hz}$ , H-15;  $\delta 1.25$ , 6H, *d*,  $J = 7\text{ Hz}$ , Me-16 and Me-17) and a singlet (3H) at  $\delta 1.34$  assigned to

the tertiary Me-18. A singlet at  $\delta 7.74$  (2H, one exchangeable with  $\text{D}_2\text{O}$ ) was ascribed to the chelated phenolic hydroxyl group and to H-20. The signal centred at  $\delta 7.5$  (1H, *dd*,  $J = 3$  and  $0.4\text{ Hz}$ ) was attributed to H-7, the  $0.4\text{ Hz}$  coupling constant indicated a long range coupling with H-20. The proton geminal to the lactone closure was responsible for a double doublet observed at  $\delta 4.73$  (1H,  $J = 11$  and  $3\text{ Hz}$ ), it was ascribed to H-6 as it was shown to be coupled to H-7 and H-5 by double resonance experiments. Irradiation at  $\delta 4.73$  (H-6) transformed the doublet at  $\delta 7.5$  (H-7) into a broad singlet and a doublet ( $J = 11\text{ Hz}$ ) at  $\delta 2.6$  (H-5) into a singlet. The coupling constant of  $11\text{ Hz}$  indicated a *trans*-diaxial relationship between H-6 and H-5; H-5 is  $\alpha$ -axially oriented on biogenetic considerations [7], therefore H-6 must be  $\beta$ -axial. A signal at  $\delta 6.65$  (*br t*,  $W_{\frac{1}{2}} = 10\text{ Hz}$ ) was ascribed to the vinylic H-1. Irradiation at  $\delta 2.5$  (H-5 and H-2) transformed the signal at  $\delta 6.65$  into a singlet and the double doublet at  $\delta 4.73$  (H-6) into a doublet ( $J = 3\text{ Hz}$ ). It also simplified a double triplet observed at  $\delta 1.85$  (1H,  $J = 13$  and  $3\text{ Hz}$ ) to a doublet ( $J = 13\text{ Hz}$ ) which could be ascribed to the  $3\beta$  proton.

The  $^{13}\text{C}$  NMR spectrum of anastomosine (Table 2) is in agreement with the structure (3) proposed for it, the assignments were made by comparison with the data of similar structures [7, 8]. C-6 was observed as a doublet at  $\delta 78.60$ .

Catalytic hydrogenation of 3 followed by air treatment, gave the tetrahydro-derivative 4, in which hydrogenation of the 1, 10 double bond occurred from the less hindered  $\alpha$  face of the molecule. In the  $^1\text{H}$  NMR of 4 (Table 1), H-6 is observed as a triple doublet at  $\delta 4.3$  ( $J = 10$  and  $4\text{ Hz}$ ) due to the coupling of this proton with the C-7 methylene (H-7 $\beta$ ,  $\delta 3.8$ , *dd*,  $J = 16$  and  $4\text{ Hz}$ ; H-7 $\alpha$ ,  $\delta 2.57$ , *dd*,  $J = 16$  and  $10\text{ Hz}$ ) and H-5. The paramagnetic displacement observed for H-7 $\beta$  ( $\delta 3.8$ ) can be attributed to the deshielding effect exerted by the C-14 carbonyl group (molecular models). These data support the assignment of stereochemistry for C-5 and C-6 proposed for anastomosine (3). The  $\beta$  proton at C-20 is also deshielded by the C-11 carbonyl group and appears at  $\delta 3.05$  as a double doublet ( $J = 14$  and  $6\text{ Hz}$ ), the  $6\text{ Hz}$  coupling constant is adequate

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Table 1  $^1\text{H}$  NMR data of compounds 3–7 ( $\text{CDCl}_3$ , TMS as int. stand.)

H	3	4	5	6	7
1	6.65 t (4)				
3	1.85 dt (13, 3)				
5	2.60 d (11)	2.08 m	2.07 m	2.75 m	2.05 m
6	4.73 dd (11, 3)	4.30 td (10, 4)	4.29 td (10, 4)	6.13 dd (11.5, 5)	4.28 td (10, 3)
7	7.50 dd (3, 0.4)	$\alpha$ 2.57 dd (16, 10)	2.55 dd (15, 10)	6.58 dd (11.5, 2)	2.78 dd (14, 10)
		$\beta$ 3.80 dd (16, 4)	3.72 dd (15, 4)		3.64 dd (14, 3)
15	3.38 sept (7)	3.23 sept (7)	3.25 sept (7)	3.24 sept (7)	3.40 sept (7)
Me-16	1.25 d (7)	1.23 d (7)	1.21 d (7)	1.20 d (7)	1.33 d (7)
Me-17				1.22 d (7)	
Me-18	1.34 s	1.30 s $\alpha$ 2.50 br d (14)	1.31 s 2.44 br d (14)	1.27 s 2.42 m	1.30 s 2.57 dd (12, 3)
20	7.74 s	$\beta$ 3.05 dd (14, 6)	3.02 dd (14, 6)	3.01 dd (12, 6)	3.14 dd (12, 6)
OH	7.74 s	7.03	—		3.75 s
OMe	—	—	3.96 s	3.93 s	3.84 s 3.64 s
$\text{CO}_2\text{Me}$	—	—	—	3.64 s	—

Coupling constants in Hz are in parentheses. Chemical shifts are in  $\delta$  values.

for a dihedral angle of H-20 $\beta$ –C-20–C-10–H-10 $\alpha$  of  $\approx 50^\circ$ , therefore H-10 must be  $\alpha$  axially oriented.

Reductive methylation of anastomosine (3) yielded three main products which were characterized by spectral means. One of them was proved to be the methyl ether (5) of the tetrahydroderivative 4. The less polar product was characterized as the methyl ester derivative 6. It showed in the IR spectrum a band at  $1722\text{ cm}^{-1}$  due to the ester moiety. In the  $^1\text{H}$  NMR spectrum (Table 1) a singlet (3H) at  $\delta 3.64$  was assigned to this function. It also showed the methyl ether bound to C-12 as a singlet (3H) at  $\delta 3.93$ . The AB portion of an ABX system was observed at  $\delta 6.58$  (dd,  $J = 11.5$  and  $2\text{ Hz}$ ) and  $6.13$  (dd,  $J = 11.5$  and  $5\text{ Hz}$ ). These signals were assigned to H-7 and H-6, respectively. The coupling constant of  $11.5\text{ Hz}$  indicated a *cis* double bond in a seven-membered ring [10]. The shape and coupling constants of H-6 and H-7 are equivalent to the proton resonance signals due to H-6 and H-7 in the spectrum of icetexone (2). The formation of product 6 supports the  $\gamma$ -lactone closure at C-6 and the relative configuration proposed for C-5 in anastomosine (3).

The third product obtained proved to have structure 7 in which the C-ring has been aromatised. The IR spectrum of 7 showed a  $\gamma$ -lactone band at  $1764\text{ cm}^{-1}$  and a band at  $1602\text{ cm}^{-1}$  due to the aromatic C ring. In the  $^1\text{H}$  NMR spectrum of 7 (Table 1), three singlets (3H each) were ascribed to the methyl ether groups. The proton geminal to the lactone closure, H-6, was observed as a

triple doublet at  $\delta 4.28$  ( $J = 10$  and  $3\text{ Hz}$ ). The assignment of the signals (Table 1) was based on double resonance experiments in a  $\text{CDCl}_3/\text{C}_6\text{D}_6$  (1:1) solution.

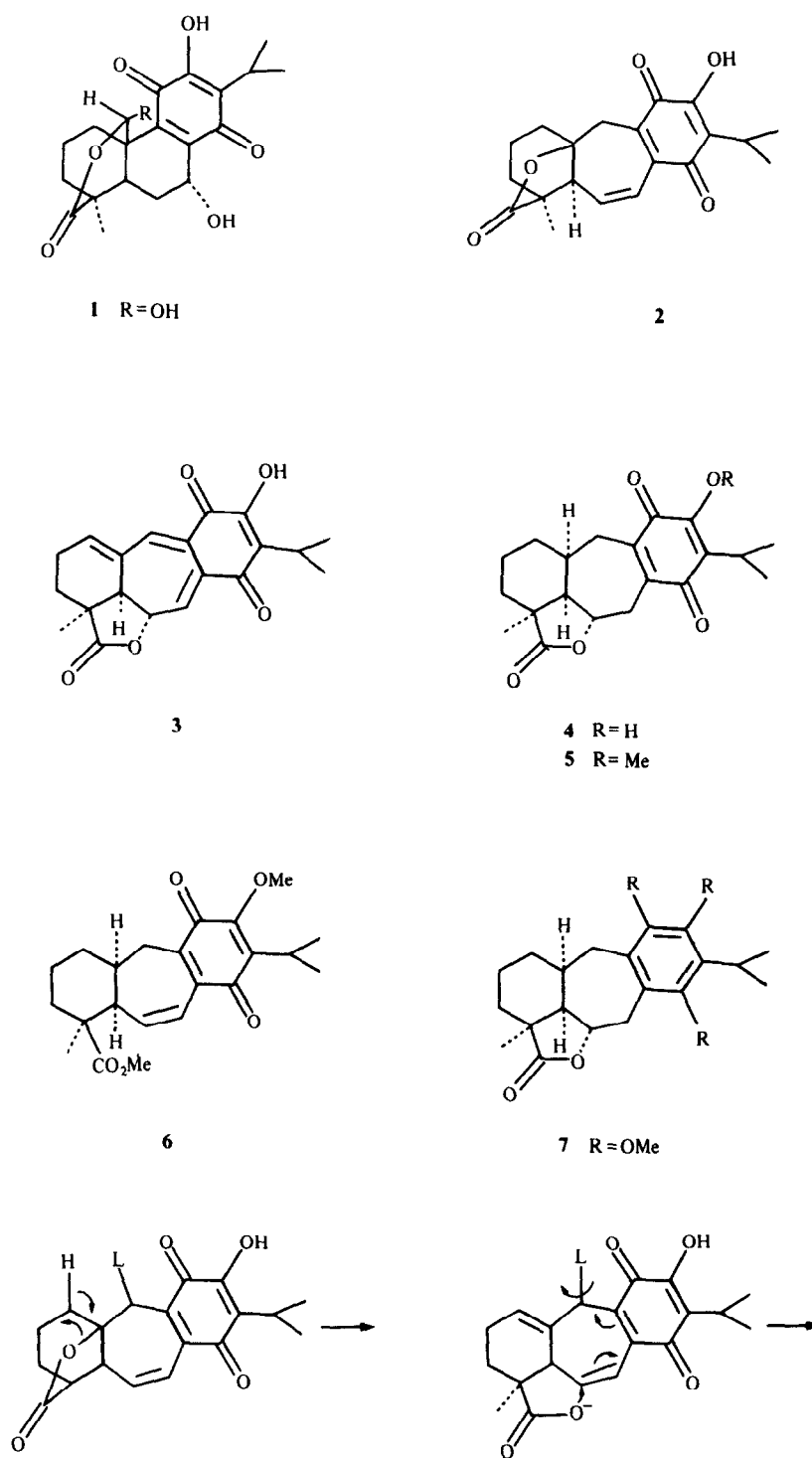
The formation of products 4–7 under reductive conditions is in accordance with structure 3 proposed for anastomosine.

Anastomosine (3) can be derived biogenetically from a 20-substituted icetexone derivative as shown in Scheme 1. The isolation of icetexone (2) and anastomosine from the same source makes this biogenetical relationship more plausible.

The presence of icetexone and anastomosine in *S. anastomosans* provides chemical support for its classification in Section *Tomentellae* (*salvia*, subgenus *Calosphace*), as this type of rearranged abietane diterpenoids have been found in all of the species of *Salvia* of this Section studied up to now [7, 8]. We propose the name of *icetexane* for the 9(10 $\rightarrow$ 20)-abeo-abietane skeleton.

#### EXPERIMENTAL

Mps uncorr. MS were obtained at  $70\text{ eV}$  by direct inlet.  $^1\text{H}$  and  $^{13}\text{C}$  NMR were performed at  $80$  and  $20\text{ MHz}$ , respectively, using TMS as int. standard. Plant material was collected at  $5\text{ km}$  from Tamazulapán on the way to Chilapa (Oaxaca, México) and a voucher specimen (MEXU 4773) is deposited at the Herbarium of the Instituto de Biología, UNAM.



*Isolation of the constituents from Salvia anastomosans.* Dried aerial parts of *S. anastomosans* Ramamoorthy, (2.6 kg) were extracted with  $\text{Me}_2\text{CO}$  (20 l) at room temp. for one week. The solvent was removed under red. pres. and the gummy residue obtained (300 g) was chromatographed over silica gel (1 kg deactivated with 10%  $\text{H}_2\text{O}$ ). Mixtures of petrol-EtOAc of increasing polarity were used as eluents. From the fractions

eluted with petrol-EtOAc (19:1) conacytone (1, 3.5 g) and icetexone (2, 558 mg) were isolated and identified by comparison with authentic samples. From the fractions eluted with petrol-EtOAc (9:1), a mixture of oleanolic and ursolic acids was obtained, which were identified as their methyl esters

Repeated chromatography of the non-crystalline fractions eluted with petrol-EtOAc (9:1) yielded 345 mg of anastomosine

Table 2  $^{13}\text{C}$  NMR chemical shifts of compound **3** (20 HMz,  $\text{CDCl}_3$ , TMS as int standard)

C	
1	141.57 $d^a$
2	23.08 $t^b$
3	25.03 $t^b$
4	41.53 $s$
5	47.72 $d$
6	78.60 $d$
7	143.06 $d$
8	131.98 $s$
9	133.79 $s$
10	124.26 $s$
11	181.39 $s$
12	155.12 $s$
13	129.16 $s$
14	182.74 $s$
15	25.37 $d$
16	19.63 $q^c$
17	19.45 $q^c$
18	21.25 $q$
19	179.73 $s$
20	140.80 $d^a$

SFORD multiplicities are in parenthesis.

<sup>a,b,c</sup> Values may be interchanged

(**3**) as a yellow crystalline product mp 207–215°,  $[\alpha]_D^{20} = +426.47$  (pyridine,  $c$  0.034), UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 200 (8000), 277 (4340), 330 (5060), IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$  3361, 1778, 1656, 1602;  $^1\text{H}$  NMR see Table 1,  $^{13}\text{C}$  NMR see Table 2; MS  $m/z$  (rel. int.) 340  $[\text{M}]^+$  (100), 312 (12), 284 (8), 269 (14).

**Catalytic hydrogenation of 3** Anastomosine (**3**, 70 mg) in EtOAc (10 ml) was hydrogenated using Pd-C (5%, 14 mg) as catalyst for 2 hr. The catalyst was removed by filtration and the residue obtained after removal of the solvent, was dissolved in  $\text{Me}_2\text{CO}$  and aerated. The crystalline product obtained (**4**) showed mp 218–220° from  $\text{CH}_2\text{Cl}_2$ –petrol, IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$  3408, 1768, 1642,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) see Table 1, MS  $m/z$  (rel. int.) 344  $[\text{M}]^+$  (100), 298 (18), 95 (40), 55 (40), 41 (46).

**Reductive methylation of 3** Anastomosine (**3**, 170 mg, 0.494 mmol) was hydrogenated at room temp in EtOAc (10 ml) using Pd-C (5%, 50 mg) as catalyst for 6 hr. After the usual work-up the product of the reaction was dissolved in  $\text{Me}_2\text{CO}$  (40 ml)

and treated with freshly dist  $(\text{MeO})_2\text{SO}_2$  (0.2 ml, 1.5 mmol) and  $\text{K}_2\text{CO}_3$  (2 g). The reaction mixture was stirred for 40 min under reflux and 16 hr at room temp.,  $(\text{MeO})_2\text{SO}_2$  (0.2 ml) and  $\text{K}_2\text{CO}_3$  (2 g) were added and the reflux continued for 2 hr. The reaction mixture was filtered, the solvent removed under red. pres. and the residue stirred with  $\text{H}_2\text{O}$  for 3 hr and extracted with EtOAc. The organic phase was washed, dried and the solvent removed. The crude product obtained was separated by flash chromatography over silica gel. The less polar product (**6**, 16.4 mg, 9%) was obtained as a yellow non-crystalline product. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 200 (7200), 265 (3000), 300 (1450), IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$  1722, 1646, 1602, 1118,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) see Table 1, MS  $m/z$  (rel. int.) 372  $[\text{M}]^+$  (100), 312 (37), 297 (27), 272 (34), 257 (18).

The second product obtained, **7** (22.4 mg, 11.68%), showed IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$  1764, 1602, 1118,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) see Table 1, MS  $m/z$  (rel. int.) 388  $[\text{M}]^+$  (100), 373 (30.4), 95 (17), 91 (16).

The most polar compound **5** was obtained as a yellow non-crystalline product (28.8 mg, 16.28%). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 200 (4200), 273 (3200), IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$  1768, 1654, 1602, 1117,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) see Table 1, MS  $m/z$  (rel. int.) 358  $[\text{M}]^+$  (100), 313 (5), 297 (8), 243 (10), 207 (6), 95 (20), 91 (13), 53 (12).

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